Chem-Impex International, Inc.

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Certificate of Analysis

32520 Catalogue Number

002917-140 **Lot Number**

Strychnine nitrate **Product**

66-32-0 **CAS Number**

 $C_{21}H_{22}N_2O_2 \bullet HNO_3$ **Molecular Formula**

Molecular Weight 397.43

White crystalline powder **Appearance**

1 gram of Strychnine Nitrate dissolve in 45ml of H2O in **Solubility**

150ml of C2H5OH,

in 50ml of Glycerin &in 105ml of Chloroform. 1 gm is suitable in 10ml of boiling water & in 80ml of C2H5OH at

60 °C.

It is insoluble in Ether

Odor Odorless **Acidity** 0.25 ml

Readily Carbonizable

Substance

Dissolve 250mg of Strychnine Nitrate in 5ml of H2SO4, the color of the solution is not deeper than a matching fluid consisting of 2.5ml of Cobalt Chloride, 2.5ml of Ferric

Chloride and 0.2ml of Cupric Sulfate

0.03% **Residue On Ignition** 99% Assay by titration

Store at RT Storage

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Additional Information

Taste: Extremely bitter

Brucine : Free Identification:

A) 0.02g of preparation is placed in a porcelain dish, 5ml of concentrated Sulfuric Acid and a few crystal of Potassium Bichromate are added. When carefully shaking of the dish violet and blue rapidly

disappearing bands are formed.

B) 0.01g of preparation is placed in porcelain dish, a few drops of

Diphenylamine are added, a blue color is produced.

C) Superimpose a solution of Strychnine Nitrate (1 in 50) upon Diphenyl test solution, contained in a test tube a Blue color develop at

the zone of contact

Chloride: Acidify 20ml of solution of Strychnine Nitrate (1 in 100), with 2 drops of dilute HNO3 added 5 drops of Silver Nitrate test solution, no procaine is produced immediately

Sulfate: Acidify the 2ml of Strychnine Nitrate solution (1 in 100) with 2 drops of HCL & added 5 drops of Barium Chloride testing solution, no

turbidity produces immediately

Remarks See 1

See material safety data sheet for additional information

For laboratory use only

The foregoing is a copy of the Certificate of Analysis as provided by our supplier

Bala Kumar

Quality Control Manager